2011 Water Quality Report

By the Numbers...

Our water quality team conducts more than 300,000 tests per year in our statecertified water quality laboratory. Scientists, chemists, and microbiologists test the water for more than 140 contaminants with equipment so sensitive it can detect levels as low as one part per billion that's like one pinch of salt in a 400-ton bag of potato chips, one sheet in a roll of toilet paper stretched three times around the equator, or three seconds in a century.



East Los Angeles District

2000 South Tubeway Avenue, Commerce, CA 90040 Customer Center: (323) 722-8601 or 24-Hour Emergencies: (323) 263-4145 infoELA@calwater.com www.calwater.com

At California Water Service Company (Cal Water), we are committed to providing a reliable, high-quality water supply to you and your family at the lowest price possible.

In this report, you'll see how your water compares to state and federal standards. We are pleased to confirm that your water met or surpassed all primary water quality standards in this reporting period.

We have also included information about how you can use water more efficiently. Although tap water costs a fraction of the price of bottled water, water rates throughout the country are increasing, due in part to increasingly strict water quality and environmental standards, infrastructure replacement needs, and higher costs for everything from electricity to labor. So using water as efficiently as possible is critical not only to ensure that we have enough water to meet your needs and reduce per capita water use by 20% by 2020 as required by state law, but also to help you control your water bill.

If you have any questions, suggestions, or concerns, please contact your local Customer Center, either by phone or through our web site. Also, please watch for bill inserts (which are also available online for customers using paperless billing), where you will find announcements about any water-related public meetings and important information about your water. Additional information and time-sensitive announcements can be found at www.calwater.com.

Dan Armendariz

District Manager East Los Angeles District

About Your Water System

Cal Water has provided high-quality water utility services in the East Los Angeles area since 1928. In addition to the customers in our East Los Angeles system, we serve customers through operating contracts with the Cities of Commerce and Montebello. To meet our customers' needs, we use a combination of local groundwater and purchased water from the Metropolitan Water District of Southern California (MWD), which is imported from the Colorado River and the State Water Project in northern California. The East Los Angeles water system currently includes 272 miles of pipeline, nine active wells, 29 booster pumps, 16 storage tanks, and three MWD connections.

We hope to complete a new 3-million-gallon storage reservoir in 2012. Cal Water proactively maintains and upgrades its facilities to ensure a reliable, high-quality supply.

Use Water Wisely. It's Essential.

You're busy. So why should you add water conservation to your mile-long to-do list? Simply put, water's got things to do, too – it keeps us clean and healthy, protects us from fire, and is used to grow our food and make the products we use every day. And there's no way to increase the earth's finite water supply. That's why we're here to help you use water wisely – because it's essential.

Under the Microscope..

All drinking water, including bottled water, may reasonably be expected to contain at least small amounts of some contaminants. The presence of contaminants does not necessarily indicate that water poses a health risk. More information about contaminants and potential health effects can be obtained by calling the United States Environmental Protection Agency (USEPA) Safe Drinking Water Hotline at (800) 426-4791.

The sources of drinking water (both tap and bottled) include rivers, lakes, streams, ponds, reservoirs, springs, and wells. As water travels over the surface of the land or through the ground, it dissolves naturally occurring minerals and, in some cases, radioactive material, and can pick up substances resulting from the presence of animals or human activity. Contaminants that may be present in source water include:

Microbial contaminants, such as viruses and bacteria, which may come from sewage treatment plants, septic systems, agricultural livestock operations, and wildlife.

Inorganic contaminants, such as salts and metals, which can be naturally occurring or result from urban stormwater runoff, industrial or domestic wastewater discharges, oil and gas production, mining, or farming.

Pesticides and herbicides, which may come from a variety of sources such as agriculture, urban stormwater runoff, and residential uses.

Organic chemical contaminants, including synthetic and volatile organic chemicals, which are byproducts of industrial processes and petroleum production, and can also come from gas stations, urban stormwater runoff, agricultural application, and septic systems.

Radioactive contaminants, which can be naturally occurring or the result of oil and gas production and mining activities.

In order to ensure that tap water is safe to drink, the USEPA and the California Department of Public Health (CDPH) prescribe regulations that limit the amount of certain contaminants in water provided by public water systems. CDPH regulations also establish limits for contaminants in bottled water, which must provide the same protection for public health.

Some people may be more vulnerable to contaminants in drinking water than the general population. Immunocompromised people, such as those with cancer undergoing chemotherapy, those who have undergone organ transplants, those with HIV/AIDS or other immune system disorders, some elderly people, and infants, can be particularly at risk from infections. These people should seek advice from their health care providers about drinking water. USEPA/Centers for Disease Control and Prevention (CDC) guidelines on appropriate means to lessen the risk of infection by Cryptosporidium and other microbial contaminants are available from the Safe Drinking Water Hotline at (800) 426-4791.

Lead in Water

If present, elevated levels of lead can cause serious health problems, especially for pregnant women and young children. Lead in drinking water comes primarily from materials and components associated with service lines and home plumbing.

The water delivered by Cal Water to your meter meets all water quality standards, but your home plumbing can affect water quality. When your water has been sitting for several hours, you can minimize the potential for lead exposure by flushing your tap for 30 seconds to two minutes before using water for drinking or cooking.

If you are concerned about lead in your water, you may wish to have your water tested by a private lab. Information on lead in drinking water, testing methods, and steps you can take to minimize exposure is available from the Safe Drinking Water Hotline or at www.epa.gov/safewater/lead.

Key Definitions

Exceeded Standard: Over a primary MCL, a secondary MCL, or an action level, as determined by the California Department of Public Health. For some compounds, compli-

ance is determined by averaging the results for one source for a year.

Maximum Contaminant Level (MCL): The highest level of a contaminant that is allowed in drinking water. Primary MCLs protect public health and are set as close to the PHGs (or MCLGs) as are economically and technologically feasible. Secondary MCLs relate to the odor, taste, and appearance of drinking water.

Maximum Contaminant Level Goal (MCLG): The level of a contaminant in drinking water below which there is no known or expected risk to health. MCLGs are set by the United States Environmental Protection Agency (USEPA).

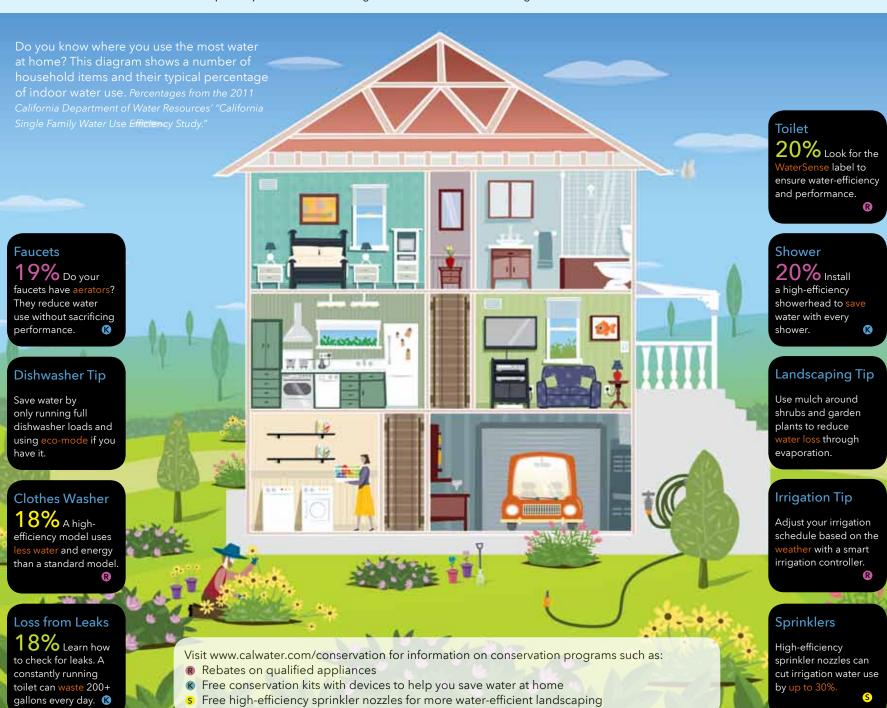
Maximum Residual Disinfectant Level (MRDL): The highest level of a disinfectant allowed in drinking water. There is convincing evidence that addition of a disinfectant is necessary for control of microbial contaminants.

Maximum Residual Disinfectant Level Goal (MRDLG): The level of a drinking water

disinfectant below which there is no known or expected risk to health. MRDLGs do not reflect the benefits of the use of disinfectants to control microbial contaminants. Notification Level (NL): A health-based advisory level for an unregulated contaminant in drinking water. It is used by the California Department of Public Health to provide guidance to drinking water systems.

Primary Drinking Water Standard (PDWS): MCLs and MRDLs for contaminants that affect health, along with their monitoring, reporting, and water treatment requirements. Public Health Goal (PHG): The level of a contaminant in drinking water below which there is no known or expected risk to health. PHGs are set by the California Environmental Protection Agency's Office of Environmental Health Hazard Assessment. Regulatory Action Level (AL): The concentration of a contaminant which, if exceeded, triggers treatment or other required action by the water provider. Treatment Technique (TT): A required process intended to reduce the level of a contami-

nant in drinking water.



How to Read This Table

Cal Water tests your water for more than 140 regulated contaminants and dozens of unregulated contaminants. A list of regulated contaminants can be found in the Water Quality section of calwater.com. This table lists only those contaminants that

In the table, water quality test results are divided into two main sections: "Primary Drinking Water Standards" and "Secondary Drinking Water Standards and Unregulated Compounds." Primary standards protect public health by limiting the levels of certain constituents in drinking water. Secondary standards are set for substances that could affect the water's taste, odor, or appearance. Selected unregulated substances (hardness and sodium, for example) are listed for your information.

 $\mu S/cm = \text{measure of}$ specific conductance

n/a = not applicable

ND = not detected

NTU = nephelometric

pCi/L = picoCuries per liter(measure of radioactivity)

ppb = parts per billion (micrograms per liter)

ppm = parts per million (milligrams per liter)

ppt = parts per trillion (nanograms per liter)

SMCL = secondary

Your Water Quality Report

Primary Drinking Wa	ator Stanc	arde				Gre	oundwater	Purchase	d Surface Water	
Radiological	Year Tested		MCL	PHC (MCIC)	Exceeded Standard?	Range	Average	Range		Source of Substance
Gross alpha particle activity	2003-2011		15		No No	ND-5.4	1.0	ND-3	Average 1	Erosion of natural deposits
Gross beta particle activity	2003-2011	pCi/L	50	(0)	No	n/a	1.0	ND-6	1.3	Decay of natural and man-made deposits
Radium 226	2010-2011		5	0.05 (0)	No	ND-1.9	0.5	ND-0		Erosion of natural deposits
Radium 228	2010-2011		5	0.05 (0)	No	ND-1.9 ND-1.1	0.03	ND ND		Erosion of natural deposits
										•
Uranium	2002-2011	•	20	0.43	No	ND-3.4	1.8	ND-2	1.6	Erosion of natural deposits
Inorganic Chemicals	Year Tested		MCL (SMCL)	, ,		Range	Average	Range	Average	Source of Substance
Aluminum	2009-2011		1 (0.2)	0.6	No	ND		ND-0.2	0.1	Erosion of natural deposits; residue from some surface water treatment processes
Arsenic ¹	2009-2011		10	0.004	No	ND-9.0	5.2	ND-2.3	0.8	Erosion of natural deposits; runoff from orchards; glass and electronics production wastes
Barium	2009-2011		1	2	No	ND-0.2	0.1	ND		Discharges of oil-drilling waste and from metal refineries; erosion of natural deposits
Chromium	2009-2011	ppb	50	(100)	No	ND-11.1	1.9	ND		Discharge from steel and pulp mills and chrome plating; erosion of natural deposits
Fluoride	2009-2011	ppm	2	1	No	0.23-0.40	0.30	n/a		$Erosion \ of \ natural \ deposits; \ water \ additive \ that \ promotes \ strong \ teeth; \ discharge \ from \ fertilizer \ and \ aluminum \ factories$
Fluoride (treatment by MWD) ²	2011	ppm	2	1	No	n/a		0.2-1.0	0.8	Water additive for dental health
Nitrate (as nitrate) ³	2011	ppm	45	45	No	1.6-26.0	18.1	ND-2.3	0.1	Runoff and leaching from fertilizer use; leaching from septic tanks and sewage; erosion of natural deposits
Perchlorate	2011	ppb	6	6	No	ND-4.7	1.4	ND		Inorganic chemical used in solid rocket propellant, fireworks, explosives, flares, matches, and a variety of industries; Usually gets into drinking water as a result of environmental contamination from historic aerospace or other industrial
Selenium	2009-2011	ppb	50	(50)	No	ND-6.1	0.8	ND		ostaning yets mist unifixing water as a result of environmental contamination normalisation decreaspace of other industrial operations that used or use, store, or dispose of perchlorate and its salts. Discharge from petroleum, glass, and metal refineries; erosion of natural deposits; discharge from mines and chemical
										manufacturers; runoff from livestock lots (feed additive)
Inorganic Chemicals	Year Tested		MCL	PHG (MCLG)			Lowest Monthly Percent			Source of Substance
Turbidity (surface water requiring filtration) ⁴	2011	NTU	TT=0.3	n/a	No	n/a		0.08	100	Soil runoff
Organic Chemicals	Year Tested	Unit	MCL	PHG (MCLG)	Exceeded Standard?	Range	Average	Range	Average	Source of Substance
1,1-Dichloroethylene	2011	ppb	6	10 (7)	No	ND-4.2	0.6	ND	•	Discharge from industrial chemical factories
Tetrachloroethylene (PCE)	2011	ppb	5	0.06	No	ND-1.7	0.6	ND		Discharge from factories, dry cleaners, and auto shops (metal degreaser)
Trichloroethylene (TCE)	2011	ppb	5	1.7	No	ND-1.2	0.4	ND		Discharge from metal-degreasing sites and other factories
memoroemyrene (rez)	2011	ppu	,	1.7	140	140-1.2	Highest Running	ND	Highest Running	Discharge from metal-degreasing sites and other factories
DBP Precursor	Year Tested	Unit	MRDL	MRDLG	Exceeded Standard?	Range	Annual Average	Range	Annual Average	Source of Substance
Total organic carbon ⁵	2011	ppm	n/a	n/a	No	0.4-0.7	0.6	1.6-3	2.2	Various natural and man-made sources
Disinfectant and Disinfection		FF					Highest Running		Highest Running	
Byproducts	Year Tested	Unit	MCL	(MCLG)	Exceeded Standard?	Range	Annual Average	Range	Annual Average	Source of Substance
Bromate	2011	ppb	10	(0)	No	n/a		ND-8.8	5.9	Byproduct of drinking water chlorination
Disinfestant and Disinfestion							Distribution	System-Wide		
Disinfectant and Disinfection Byproducts	Year Tested	Unit	MCL	PHG (MCIG)	Exceeded Standard?		Range	•	nning Annual Average	Source of Substance
Chloramine	2011	ppm	4	4	No		0.5-3.1	riigiiostiia	1.9	Drinking water disinfectant added for treatment
Total haloacetic acids ⁶	2011	pph	60	n/a	No		ND-61.3		17.9	Byproduct of drinking water chlorination
Total Haloacetic acids	2011	ppp	00	11/4	140		140 01:5		17.7	byproduct or drinking water enformation
Total tribalomothanocó	2011	nnh	80	n/a	No		ND_84 A		27.7	Ruproduct of drinking water chlorination
Total trihalomethanes ⁶	2011	ppb	80	n/a	No		ND-84.6	Custom Mi	27.7	Byproduct of drinking water chlorination
Other Regulated Sul	bstances					00	Distribution		de	
Other Regulated Sul Metals	bstances Year Tested	Unit	AL	PHG	Exceeded Standard?	90	Distribution Oth Percentile		d e / Total # Sites Sampled	Source of Substance
Other Regulated Sul Metals Copper	bstances Year Tested 2011	Unit ppm	AL 1.3	PHG 0.3	Exceeded Standard? No		Distribution Oth Percentile 0.014	# Sites > AL	de / Total # Sites Sampled 0/50	
Other Regulated Sul Metals Copper Secondary Drinking	Year Tested 2011 Water Sta	Unit ppm andare	AL 1.3 ds and Un	PHG 0.3 rregulated	Exceeded Standard? No Compounds	Gro	Distribution Oth Percentile 0.014 Dundwater	# Sites > AL	de /Total # Sites Sampled 0/50 d Surface Water	Source of Substance Internal corrosion of household plumbing systems; erosion of natural deposits; leaching from wood preservatives
Other Regulated Suit Metals Copper Secondary Drinking Inorganic Chemicals	Year Tested 2011 Water Sta Year Tested	Unit ppm andard Unit	AL 1.3 ds and Un SMCL	PHG 0.3 nregulated PHG (MCLG)	Exceeded Standard? No Compounds Exceeded Standard?	Gro Range	Distribution Oth Percentile 0.014 Dundwater Average	# Sites > AL Purchase Range	de /Total # Sites Sampled 0/50 d Surface Water Average	Source of Substance Internal corrosion of household plumbing systems; erosion of natural deposits; leaching from wood preservatives Source of Substance
Other Regulated Sul Metals Copper Secondary Drinking Inorganic Chemicals Boron	Year Tested 2011 Water Sta Year Tested 2011	Unit ppm andard Unit ppm	AL 1.3 ds and Un SMCL NL=1	PHG 0.3 nregulated PHG (MCLG) n/a	Exceeded Standard? No Compounds Exceeded Standard? No	Gro Range 0.2-0.3	Distribution Oth Percentile 0.014 Oundwater Average 0.25	# Sites > AL Purchase Range 130-190	de / Total # Sites Sampled 0/50 d Surface Water Average 150	Source of Substance Internal corrosion of household plumbing systems; erosion of natural deposits; leaching from wood preservatives Source of Substance Erosion of natural deposits
Other Regulated Sul Metals Copper Secondary Drinking Inorganic Chemicals Boron Calcium	Year Tested 2011 Water Sta Year Tested 2011 2009-2011	Unit ppm andaro Unit ppm ppm	AL 1.3 ds and Un SMCL NL=1 n/a	PHG 0.3 nregulated PHG (MCLG) n/a n/a	Exceeded Standard? No Compounds Exceeded Standard? No No	Gro Range 0.2-0.3 60-100	Distribution Oth Percentile 0.014 Dundwater Average 0.25 82	# Sites > AL Purchase Range 130-190 26-55	de /Total # Sites Sampled 0/50 d Surface Water Average 150 42	Source of Substance Internal corrosion of household plumbing systems; erosion of natural deposits; leaching from wood preservatives Source of Substance Erosion of natural deposits Erosion of natural deposits
Other Regulated Sul Metals Copper Secondary Drinking Inorganic Chemicals Boron Calcium Chloride	Year Tested 2011 Water Sta Year Tested 2011 2009-2011 2009-2011	Unit ppm andare Unit ppm ppm	AL 1.3 ds and Un SMCL NL=1 n/a 500	PHG 0.3 nregulated PHG (MCLG) n/a n/a n/a	Exceeded Standard? No Compounds Exceeded Standard? No No No	Range 0.2-0.3 60-100 62-160	Distribution Oth Percentile 0.014 Dundwater Average 0.25 82 74	# Sites > AL Purchase Range 130-190 26-55 59-76	de /Total # Sites Sampled 0/50 d Surface Water Average 150 42 69	Source of Substance Internal corrosion of household plumbing systems; erosion of natural deposits; leaching from wood preservatives Source of Substance Erosion of natural deposits Erosion of natural deposits Erosion of natural deposits; seawater influence
Other Regulated Sul Metals Copper Secondary Drinking Inorganic Chemicals Boron Calcium Chloride Chromium 6+	Year Tested 2011 Water Sta Year Tested 2011 2009-2011 2009-2011 2011	Unit ppm andard Unit ppm ppm ppm	AL 1.3 ds and Un SMCL NL=1 n/a 500 n/a	PHG 0.3 nregulated PHG (MCLG) n/a n/a n/a	Exceeded Standard? No Compounds Exceeded Standard? No No No No	Range 0.2-0.3 60-100 62-160 ND-7.2	Distribution Oth Percentile 0.014 Dundwater Average 0.25 82 74 1.7	# Sites > AL Purchase Range 130-190 26-55 59-76 0.09-0.2	de /Total # Sites Sampled 0/50 d Surface Water Average 150 42 69 0.13	Source of Substance Internal corrosion of household plumbing systems; erosion of natural deposits; leaching from wood preservatives Source of Substance Erosion of natural deposits Erosion of natural deposits Erosion of natural deposits Erosion of natural deposits; seawater influence Discharge from steel and pulp mills and chrome plating; erosion of natural deposits
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Other Regulated Sul Metals Copper Secondary Drinking Inorganic Chemicals Boron Calcium Chloride Chromium 6+ Color Hardness	Year Tested 2011 Water Sta Year Tested 2011 2009-2011 2009-2011 2009-2011 2009-2011	Unit ppm and are Unit ppm ppm ppm ppb Units ppm	AL 1.3 ds and Un SMCL NL=1 n/a 500 n/a 15 n/a	PHG 0.3 nregulated PHG (MCLG) n/a n/a n/a n/a n/a	Exceeded Standard? No Compounds Exceeded Standard? No No No No No No No No	Range 0.2-0.3 60-100 62-160 ND-7.2 ND-5 210-390	Distribution Oth Percentile 0.014 Dundwater Average 0.25 82 74 1.7 0.5 293	# Sites > AL Purchase Range 130-190 26-55 59-76 0.09-0.2 1-2 57-270	de //Total # Sites Sampled 0/50 d Surface Water	Source of Substance Internal corrosion of household plumbing systems; erosion of natural deposits; leaching from wood preservatives Source of Substance Erosion of natural deposits Erosion of natural deposits Erosion of natural deposits Erosion of natural deposits; seawater influence Discharge from steel and pulp mills and chrome plating; erosion of natural deposits Naturally occurring organic matter Erosion of natural deposits
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Other Regulated Sul Metals Copper Secondary Drinking Inorganic Chemicals Boron Calcium Chloride Chromium 6+ Color Hardness Iron ⁷ Magnesium Manganese ⁸ Odor PH Sodium	bstances Year Tested 2011 Water Sta Year Tested 2011 2009-2011 2009-2011 2009-2011 2009-2011 2009-2011 2009-2011 2009-2011 2009-2011 2009-2011 2011 2011	Unit ppm and are Unit ppm ppm ppb Units ppm ppb Units ppm ppb Units units ppm ppb Units Units ppm	AL 1.3 Is and Un SMCL NL=1 n/a 500 n/a 300 n/a 50 3 n/a n/a n/a	PHG 0.3 PHG (MCLG) n/a n/a n/a n/a n/a n/a n/a n/	Exceeded Standard? No Compounds Exceeded Standard? No	Range 0.2-0.3 60-100 62-160 ND-7.2 ND-5 210-390 ND-680 13-32 ND-57 ND-3 6.8-8.0 54-75	Distribution Oth Percentile 0.014 Dundwater Average 0.25 82 74 1.7 0.5 293 27 21 3.2 1 7.3 64	# Sites > AL Purchase Range 130-190 26-55 59-76 0.09-0.2 1-2 57-270 ND 12-21 ND ND 7-8.8 52-77	de /Total # Sites Sampled 0/50 d Surface Water Average 150 42 69 0.13 1.3 1.57	Source of Substance Internal corrosion of household plumbing systems; erosion of natural deposits; leaching from wood preservatives Source of Substance Erosion of natural deposits Erosion of natural deposits Erosion of natural deposits; seawater influence Discharge from steel and pulp mills and chrome plating; erosion of natural deposits Naturally occurring organic matter Erosion of natural deposits; industrial wastes Erosion of natural deposits Leaching from natural deposits Erosion of natural deposits Substance Leaching from natural deposits Leaching from natural deposits Leaching from natural deposits Substance Leaching from natural deposits Substance Leaching from natural deposits
Other Regulated Sul Metals Copper Secondary Drinking Inorganic Chemicals Boron Calcium Chloride Chromium 6+ Color Hardness Irron ⁷ Magnesium Manganese ⁸ Odor pH Sodium Specific conductance	bstances Year Tested 2011 Water Sta Year Tested 2011 2009-2011 2009-2011 2009-2011 2009-2011 2009-2011 2009-2011 2009-2011 2009-2011 2009-2011 2009-2011 2009-2011 2009-2011	Unit ppm ppm ppb Units units ppm ppb Units units ppm ppb Units Units ppm ppm ppm ppm ppm ppm ppm ppm ppm pp	AL 1.3 1.3 SMCL NL=1 n/a 500 n/a 15 n/a 300 n/a 50 3 n/a	PHG 0.3 PHG (MCLG) n/a n/a n/a n/a n/a n/a n/a n/	Exceeded Standard? No Compounds Exceeded Standard? No	Range 0.2-0.3 60-100 62-160 ND-7.2 ND-5 210-390 ND-680 13-32 ND-57 ND-3 6.8-8.0 54-75 720-1100	Distribution Oth Percentile 0.014 Dundwater Average 0.25 82 74 1.7 0.5 293 27 21 3.2 1 7.3 64 816	# Sites > AL Purchase Range 130-190 26-55 59-76 0.09-0.2 1-2 57-270 ND 12-21 ND ND 7-8.8 52-77 320-960	de /Total # Sites Sampled 0/50 d Surface Water Average 150 42 69 0.13 1.3 157 17	Source of Substance Internal corrosion of household plumbing systems; erosion of natural deposits; leaching from wood preservatives Source of Substance Erosion of natural deposits Erosion of natural deposits Erosion of natural deposits; seawater influence Discharge from steel and pulp mills and chrome plating; erosion of natural deposits Naturally occurring organic matter Erosion of natural deposits Leaching from natural deposits Erosion of natural deposits Leaching from natural deposits Source of Substance Erosion of natural deposits Leaching from natural deposits Leaching from natural deposits Naturally occurring organic matter Inherent characteristic of water Erosion of natural deposits; seawater influence Erosion of natural deposits; seawater influence
Other Regulated Sul Metals Copper Secondary Drinking Inorganic Chemicals Boron Calcium Chloride Chromium 6+ Color Hardness Iron ⁷ Magnesium Manganese ⁸ Odor pH Sodium Specific conductance Sulfate	bstances Year Tested 2011 Water Sta Year Tested 2011 2009-2011 2009-2011 2009-2011 2009-2011 2009-2011 2009-2011 2009-2011 2009-2011 2009-2011 2011 2009-2011 2011 2009-2011 2009-2011 2009-2011 2009-2011 2009-2011	Unit ppm ppm ppb Units ppm ppm ppm ppm ppm ppm ppm ppm ppm pp	AL 1.3 SMCL NL=1 n/a 500 n/a 15 n/a 300 n/a 50 3 n/a n/a 1600 500	PHG 0.3 PHG (MCLG) n/a n/a n/a n/a n/a n/a n/a n/	Exceeded Standard? No Compounds Exceeded Standard? No	Range 0.2-0.3 60-100 62-160 ND-7.2 ND-5 210-390 ND-680 13-32 ND-57 ND-3 6.8-8.0 54-75 720-1100 79-120	Distribution Oth Percentile 0.014 Dundwater Average 0.25 82 74 1.7 0.5 293 27 21 3.2 1 7.3 64 816 107	# Sites > AL Purchase Range 130-190 26-55 59-76 0.09-0.2 1-2 57-270 ND 12-21 ND ND 7-8.8 52-77 320-960 54-170	de /Total # Sites Sampled 0/50 d Surface Water Average 150 42 69 0.13 1.3 157	Source of Substance Internal corrosion of household plumbing systems; erosion of natural deposits; leaching from wood preservatives Source of Substance Erosion of natural deposits Erosion of natural deposits Erosion of natural deposits; seawater influence Discharge from steel and pulp mills and chrome plating; erosion of natural deposits Naturally occurring organic matter Erosion of natural deposits Leaching from natural deposits Erosion of natural deposits Leaching from natural deposits Erosion of natural deposits Erosion of natural deposits Erosion of natural deposits Erosion of natural deposits, seawater influence Erosion of natural deposits; seawater influence Erosion of natural deposits; seawater influence Erosion of natural deposits; industrial wastes
Other Regulated Sul Metals Copper Secondary Drinking Inorganic Chemicals Boron Calcium Chloride Chromium 6+ Color Hardness Iron ⁷ Magnesium Manganese ⁸ Odor pH Sodium Specific conductance Sulfate Total dissolved solids	bstances Year Tested 2011 Water Sta Year Tested 2011 2009-2011 2011 2009-2011 2009-2011 2009-2011 2009-2011 2009-2011 2009-2011 2009-2011 2011 2009-2011 2011 2009-2011 2009-2011 2009-2011 2009-2011 2009-2011 2009-2011 2009-2011	Unit ppm ppm ppb Units ppm ppm ppm ppm ppm ppm ppm ppm ppm pp	AL 1.3 SMCL NL=1 n/a 500 n/a 15 n/a 300 n/a 50 3 n/a 1600 500 1000	PHG 0.3 PHG (MCLG) n/a n/a n/a n/a n/a n/a n/a n/	Exceeded Standard? No Compounds Exceeded Standard? No	Range 0.2-0.3 60-100 62-160 ND-7.2 ND-5 210-390 ND-680 13-32 ND-57 ND-3 6.8-8.0 54-75 720-1100 79-120 440-630	Distribution Oth Percentile 0.014 Dundwater Average 0.25 82 74 1.7 0.5 293 27 21 3.2 1 7.3 64 816 107 522	# Sites > AL Purchase Range 130-190 26-55 59-76 0.09-0.2 1-2 57-270 ND 12-21 ND ND 7-8.8 52-77 320-960 54-170 280-490	de /Total # Sites Sampled 0/50 d Surface Water Average 150 42 69 0.13 1.3 157 17 8.1 65 607 122 397	Source of Substance Internal corrosion of household plumbing systems; erosion of natural deposits; leaching from wood preservatives Source of Substance Erosion of natural deposits Erosion of natural deposits Erosion of natural deposits; seawater influence Discharge from steel and pulp mills and chrome plating; erosion of natural deposits Naturally occurring organic matter Erosion of natural deposits Leaching from natural deposits; industrial wastes Erosion of natural deposits Leaching from natural deposits Naturally occurring organic matter Inherent characteristic of water Erosion of natural deposits; seawater influence Runoff/leaching from natural deposits; industrial wastes Runoff/leaching from natural deposits; industrial wastes
Other Regulated Sul Metals Copper Secondary Drinking Inorganic Chemicals Boron Calcium Chloride Chromium 6+ Color Hardness Iron ⁷ Magnesium Manganese ⁸ Odor pH Sodium Specific conductance Sulfate Total dissolved solids Turbidity (groundwater)	bstances Year Tested 2011 Water Sta Year Tested 2011 2009-2011 2011 2009-2011 2009-2011 2009-2011 2009-2011 2009-2011 2011 2009-2011 2011 2009-2011 2011 2009-2011 2009-2011 2009-2011 2009-2011 2009-2011 2009-2011 2009-2011	Unit ppm unit ppm ppm ppb Units ppm ppb Units ppm ppb Units ppm ppm ppb Units ppm ppm ppb Units Units ppm ppm ppm NTU	AL 1.3 SMCL NL=1 n/a 500 n/a 15 n/a 300 n/a 50 3 n/a 1600 500 1000 5	PHG 0.3 PHG (MCLG) n/a n/a n/a n/a n/a n/a n/a n/	Exceeded Standard? No Compounds Exceeded Standard? No	Range 0.2-0.3 60-100 62-160 ND-7.2 ND-5 210-390 ND-680 13-32 ND-57 ND-3 6.8-8.0 54-75 720-1100 79-120 440-630 ND-2.2	Distribution Oth Percentile 0.014 Dundwater Average 0.25 82 74 1.7 0.5 293 27 21 3.2 1 7.3 64 816 107 522 0.3	# Sites > AL Purchase Range 130-190 26-55 59-76 0.09-0.2 1-2 57-270 ND 12-21 ND ND 7-8.8 52-77 320-960 54-170 280-490 0.02-0.25	de /Total # Sites Sampled 0/50 d Surface Water Average 150 42 69 0.13 1.3 157 17 48.1 65 607 122 397 0.04	Source of Substance Internal corrosion of household plumbing systems; erosion of natural deposits; leaching from wood preservatives Source of Substance Erosion of natural deposits Erosion of natural deposits Erosion of natural deposits; seawater influence Discharge from steel and pulp mills and chrome plating; erosion of natural deposits Naturally occurring organic matter Erosion of natural deposits Leaching from natural deposits; industrial wastes Erosion of natural deposits Leaching from natural deposits Naturally occurring organic matter Inherent characteristic of water Erosion of natural deposits; seawater influence Runoff/leaching from natural deposits; industrial wastes Runoff/leaching from natural deposits
Other Regulated Sul Metals Copper Secondary Drinking Inorganic Chemicals Boron Calcium Chloride Chromium 6+ Color Hardness Iron ⁷ Magnesium Manganese ⁸ Odor pH Sodium Specific conductance Sulfate Total dissolved solids Turbidity (groundwater) Vanadium	bstances Year Tested 2011 Water Sta Year Tested 2011 2009-2011 2011 2009-2011 2009-2011 2009-2011 2009-2011 2009-2011 2011 2009-2011 2011 2009-2011 2009-2011 2011 2009-2011 2009-2011 2009-2011 2009-2011 2009-2011 2009-2011	Unit ppm unit ppm ppm ppb Units unit unit ppm ppm ppm ppm ppm ppm ppm NTU ppb	AL 1.3 1.3 Is and Un SMCL NL=1 n/a 500 n/a 15 n/a 300 n/a 50 3 n/a 1600 500 1000 5 NL=50	PHG 0.3 PHG (MCLG) n/a n/a n/a n/a n/a n/a n/a n/	Exceeded Standard? No Compounds Exceeded Standard? No	Range 0.2-0.3 60-100 62-160 ND-7.2 ND-5 210-390 ND-680 13-32 ND-57 ND-3 6.8-8.0 54-75 720-1100 79-120 440-630	Distribution Oth Percentile 0.014 Dundwater Average 0.25 82 74 1.7 0.5 293 27 21 3.2 1 7.3 64 816 107 522	# Sites > AL Purchase Range 130-190 26-55 59-76 0.09-0.2 1-2 57-270 ND 12-21 ND ND 7-8.8 52-77 320-960 54-170 280-490 0.02-0.25 ND-3.4	de /Total # Sites Sampled 0/50 d Surface Water Average 150 42 69 0.13 1.3 157 17 48.1 65 607 122 397 0.04 1.1	Source of Substance Internal corrosion of household plumbing systems; erosion of natural deposits; leaching from wood preservatives Source of Substance Erosion of natural deposits Erosion of natural deposits Erosion of natural deposits; seawater influence Discharge from steel and pulp mills and chrome plating; erosion of natural deposits Naturally occurring organic matter Erosion of natural deposits Leaching from natural deposits Leaching from natural deposits Leaching from natural deposits Leaching from natural deposits Erosion of natural deposits; seawater influence Erosion of natural deposits; seawater influence Erosion of natural deposits; seawater influence Runoff/leaching from natural deposits; industrial wastes Runoff/leaching from natural deposits; industrial wastes Soil runoff Erosion of natural deposits; manufacturing of alloys and steel
Other Regulated Sul Metals Copper Secondary Drinking Inorganic Chemicals Boron Calcium Chloride Chromium 6+ Color Hardness Iron ⁷ Magnesium Manganese ⁸ Odor pH Sodium Specific conductance Sulfate Total dissolved solids Turbidity (groundwater) Vanadium Disinfection Byproducts	bstances Year Tested 2011 Water Sta Year Tested 2011 2009-2011	Unit ppm Unit ppm ppm ppm ppm ppb Units ppm ppb Units units ppm ppb Units Units ppm ppb Units Units ppm ppb Units Units ppm Units Units ppm ppb Units Units	AL 1.3 is and Un SMCL NL=1 n/a 500 n/a 15 n/a 300 n/a 50 3 n/a 1600 500 1000 5 NL=50 MCL	PHG 0.3 PHG (MCLG) n/a n/a n/a n/a n/a n/a n/a n/	Exceeded Standard? No Compounds Exceeded Standard? No	Range 0.2-0.3 60-100 62-160 ND-7.2 ND-5 210-390 ND-680 13-32 ND-57 ND-3 6.8-8.0 54-75 720-1100 79-120 440-630 ND-2.2	Distribution Oth Percentile 0.014 Dundwater Average 0.25 82 74 1.7 0.5 293 27 21 3.2 1 7.3 64 816 107 522 0.3	# Sites > AL Purchase Range 130-190 26-55 59-76 0.09-0.2 1-2 57-270 ND 12-21 ND ND 7-8.8 52-77 320-960 54-170 280-490 0.02-0.25 ND-3.4 Range	de /Total # Sites Sampled 0/50 d Surface Water Average 150 42 69 0.13 1.3 157 17 8.1 65 607 122 397 0.04 1.1 Highest Annual Average	Source of Substance Internal corrosion of household plumbing systems; erosion of natural deposits; leaching from wood preservatives Source of Substance Erosion of natural deposits Erosion of natural deposits Erosion of natural deposits; seawater influence Discharge from steel and pulp mills and chrome plating; erosion of natural deposits Naturally occurring organic matter Erosion of natural deposits Leaching from natural deposits Leaching from natural deposits Leaching from natural deposits Inherent characteristic of water Erosion of natural deposits; seawater influence Runoff/leaching from natural deposits; industrial wastes Soil runoff Erosion of natural deposits; manufacturing of alloys and steel Source of Substance
Other Regulated Sul Metals Copper Secondary Drinking Inorganic Chemicals Boron Calcium Chloride Chromium 6+ Color Hardness Iron ⁷ Magnesium Manganese ⁸ Odor pH Sodium Specific conductance Sulfate Total dissolved solids Turbidity (groundwater) Vanadium Disinfection Byproducts Chlorate	bstances Year Tested 2011 Water Sta Year Tested 2011 2009-2011 2009-2011 2009-2011 2009-2011 2009-2011 2011 2009-2011 2009-2011 2009-2011 2009-2011 2009-2011 2009-2011 2009-2011 2009-2011 2009-2011 2009-2011 2009-2011 2009-2011 2009-2011 2009-2011 2011 Year Tested 2011	Unit ppm Unit ppm ppm ppm ppm ppb Units ppm ppb Units units ppm ppb Units units ppm ppm ppm ppm ppm ppm ppm ppm	AL 1.3 is and Un SMCL NL=1 n/a 500 n/a 15 n/a 300 n/a 50 3 n/a 1600 500 1000 5 NL=50 MCL NL=800	PHG 0.3 PHG (MCLG) n/a n/a n/a n/a n/a n/a n/a n/	Exceeded Standard? No Compounds Exceeded Standard? No	Range 0.2-0.3 60-100 62-160 ND-7.2 ND-5 210-390 ND-680 13-32 ND-57 ND-3 6.8-8.0 54-75 720-1100 79-120 440-630 ND-2.2	Distribution Oth Percentile 0.014 Dundwater Average 0.25 82 74 1.7 0.5 293 27 21 3.2 1 7.3 64 816 107 522 0.3	# Sites > AL Purchase Range 130-190 26-55 59-76 0.09-0.2 1-2 57-270 ND 12-21 ND ND 7-8.8 52-77 320-960 54-170 280-490 0.02-0.25 ND-3.4 Range ND-58	de /Total # Sites Sampled 0/50 d Surface Water Average 150 42 69 0.13 1.3 157 17 8.1 65 607 122 397 0.04 1.1 Highest Annual Average	Source of Substance Internal corrosion of household plumbing systems; erosion of natural deposits; leaching from wood preservatives Source of Substance Erosion of natural deposits Erosion of natural deposits Erosion of natural deposits; seawater influence Discharge from steel and pulp mills and chrome plating; erosion of natural deposits Naturally occurring organic matter Erosion of natural deposits Leaching from natural deposits Erosion of natural deposits Leaching from natural deposits Ruturally occurring organic matter Inherent characteristic of water Erosion of natural deposits; seawater influence Erosion of natural deposits; seawater influence Erosion of natural deposits; industrial wastes Runoff/leaching from natural deposits Soil runoff Erosion of natural deposits; manufacturing of alloys and steel Source of Substance Byproduct of drinking water chlorination
Other Regulated Sul Metals Copper Secondary Drinking Inorganic Chemicals Boron Calcium Chloride Chromium 6+ Color Hardness Iron ⁷ Magnesium Manganese ⁸ Odor pH Sodium Specific conductance Sulfate Total dissolved solids Turbidity (groundwater) Vanadium Disinfection Byproducts	bstances Year Tested 2011 Water Sta Year Tested 2011 2009-2011	Unit ppm Unit ppm ppm ppm ppm ppb Units ppm ppb Units units ppm ppb Units Units ppm ppb Units Units ppm ppb Units Units ppm Units Units ppm ppb Units Units	AL 1.3 is and Un SMCL NL=1 n/a 500 n/a 15 n/a 300 n/a 50 3 n/a 1600 500 1000 5 NL=50 MCL	PHG 0.3 PHG (MCLG) n/a n/a n/a n/a n/a n/a n/a n/	Exceeded Standard? No Compounds Exceeded Standard? No	Range 0.2-0.3 60-100 62-160 ND-7.2 ND-5 210-390 ND-680 13-32 ND-57 ND-3 6.8-8.0 54-75 720-1100 79-120 440-630 ND-2.2	Distribution Oth Percentile 0.014 Dundwater Average 0.25 82 74 1.7 0.5 293 27 21 3.2 1 7.3 64 816 107 522 0.3 6.0	# Sites > AL Purchase Range 130-190 26-55 59-76 0.09-0.2 1-2 57-270 ND 12-21 ND ND 7-8.8 52-77 320-960 54-170 280-490 0.02-0.25 ND-3.4 Range ND-58 ND-8	de /Total # Sites Sampled 0/50 d Surface Water Average 150 42 69 0.13 1.3 157 17 8.1 65 607 122 397 0.04 1.1 Highest Annual Average	Source of Substance Internal corrosion of household plumbing systems; erosion of natural deposits; leaching from wood preservatives Source of Substance Erosion of natural deposits Erosion of natural deposits Erosion of natural deposits; seawater influence Discharge from steel and pulp mills and chrome plating; erosion of natural deposits Naturally occurring organic matter Erosion of natural deposits Leaching from natural deposits Leaching from natural deposits Leaching from natural deposits Naturally occurring organic matter Inherent characteristic of water Erosion of natural deposits; seawater influence Runoff/leaching from natural deposits; industrial wastes Runoff/leaching from natural deposits; industrial wastes Soil runoff Erosion of natural deposits; manufacturing of alloys and steel Source of Substance
Other Regulated Sul Metals Copper Secondary Drinking Inorganic Chemicals Boron Calcium Chloride Chromium 6+ Color Hardness Iron ⁷ Magnesium Manganese ⁸ Odor pH Sodium Specific conductance Sulfate Total dissolved solids Turbidity (groundwater) Vanadium Disinfection Byproducts Chlorate n-Nitrosodimethylamine	bstances Year Tested 2011 Water Ste Year Tested 2011 2009-2011 2009-2011 2009-2011 2009-2011 2009-2011 2009-2011 2009-2011 2011 2009-2011 2009-2011 2009-2011 2009-2011 2009-2011 2009-2011 2009-2011 2009-2011 2009-2011 2009-2011 2009-2011 2009-2011 2009-2011 2009-2011 2009-2011 2009-2011 2009-2011 2009-2011 2011 Year Tested 2011	Unit ppm Unit ppm ppm ppb Units Units ppm ppb Units Units ppm ppm Units Units ppm ppm ppm ppm ppm ppm ppm ppm ppm pp	AL 1.3 is and Un SMCL NL=1 n/a 500 n/a 15 n/a 300 n/a 50 3 n/a 1600 500 1000 5 NL=50 MCL NL=800 NL=800 NL=10	PHG 0.3 PHG (MCLG) n/a n/a n/a n/a n/a n/a n/a n/	Exceeded Standard? No Compounds Exceeded Standard? No	Range 0.2-0.3 60-100 62-160 ND-7.2 ND-5 210-390 ND-680 13-32 ND-57 ND-3 6.8-8.0 54-75 720-1100 79-120 440-630 ND-2.2	Distribution Oth Percentile 0.014 Dundwater Average 0.25 82 74 1.7 0.5 293 27 21 3.2 1 7.3 64 816 107 522 0.3 6.0	# Sites > AL Purchase Range 130-190 26-55 59-76 0.09-0.2 1-2 57-270 ND 12-21 ND ND 7-8.8 52-77 320-960 54-170 280-490 0.02-0.25 ND-3.4 Range ND-58 ND-8 System-Wide	de /Total # Sites Sampled 0/50 d Surface Water Average 150 42 69 0.13 1.3 157 17 8.1 65 607 122 397 0.04 1.1 Highest Annual Average 39 3	Source of Substance Internal corrosion of household plumbing systems; erosion of natural deposits; leaching from wood preservatives Source of Substance Erosion of natural deposits Erosion of natural deposits Erosion of natural deposits; seawater influence Discharge from steel and pulp mills and chrome plating; erosion of natural deposits Naturally occurring organic matter Erosion of natural deposits Leaching from natural deposits Leaching from natural deposits Leaching from natural deposits Naturally occurring organic matter Inherent characteristic of water Erosion of natural deposits; seawater influence Erosion of natural deposits; seawater influence Erosion of natural deposits; seawater influence Runoff/leaching from natural deposits; industrial wastes Runoff/leaching from natural deposits; oli untoff Erosion of natural deposits; manufacturing of alloys and steel Source of Substance Byproduct of drinking water chlorination; industrial processes
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1 While your drinking water meets the federal and state standards for arsenic, it does contain low levels of arsenic. The arsenic standards balance the current understanding of arsenic's possible health effects against the costs of removing arsenic from drinking water. The USEPA continues to research the health effects of low levels of arsenic, which is a mineral known to cause cancer in humans at high concentrations and is linked to other health effects, such as skin damage and circulatory problems.

2 Cal Water does not add fluoride to its groundwater supply; however, low levels of fluoride occur naturally. In November 2007, Metropolitan Water District of Southern California (MWD) began fluoridating its treated surface water, which Cal Water purchases. The range of fluoride concentrations indicated under "Purchased Surface Water" is the result of samples collected from the effluent of MWD treatment plants after fluoride was added. Since the system receives a blend of groundwater with naturally occurring fluoride and fluoridated surface water, fluoride levels are checked throughout the distribution system every month to verify the actual levels at various locations. The optimal fluoride level for the East Los Angeles system is 0.8 ppm, with a control range of 0.7–1.3 ppm.

3 The average nitrate level was 18.1 ppm, with a maximum level of 26 ppm. We are closely monitoring the nitrate levels. Nitrate in drinking water at levels above 45 ppm is a health risk for infants of less than six months of age. Such nitrate levels in drinking water can interfere with the capacity of the infant's blood to carry oxygen, resulting in a serious illness; symptoms include shortness of breath and blueness of the skin. Nitrate levels above 45 ppm may also affect the ability of the blood to carry oxygen in other individuals, such as pregnant women and those with certain specific enzyme deficiencies. If you are caring for an infant or you are pregnant, you should seek advice from your health care provider.

4 For surface water systems, the treatment technique dictates that the turbidity level of the filtered water be less than or equal to 0.3 NTU in 95% of the measurements taken each month and shall not exceed 1 NTU at any time. Turbidity is a measurement of the cloudiness of water. We monitor it because it is a good indicator of the effectiveness of our filtration system.

5 Total organic carbon (TOC) has no health effects; however, TOC provides a medium for the formation of disinfection byproducts. These byproducts include trihalomethanes (THMs) and haloacetic acids (HAAs). Drinking water containing these byproducts in excess of the MCL may lead to adverse health effects such as liver, kidney, or nervous system problems, and may lead to an increased risk of cancer. Concerns regarding disinfection byproducts are based upon exposure over many years.

6 There was one sample that exceeded the MCL for total haloacetic acids and total trihalomethane acids. Source water sometimes contains high levels of total organic carbon, which reacts with chlorine used to treat filtered water to ensure that it is safe to drink, and produces disinfection byproducts. Normally, our treatment process effectively limits this formation; however, the exceptionally high levels of organic matter experienced during early 2011 allowed byproducts in one sample to form above the MCL value. Exceeding an MCL value does not result in exceeding the standard because compliance is determined by the average of samples collected. Some people who drink water containing haloacetic acids or trihalomethane acids in excess of the MCL over many years may have an increased risk of getting cancer.

7 One of 59 iron samples exceeded the secondary maximum contaminant level value (SMCL) of 300 ppb; the standard was not exceeded because compliance is based on the average of samples collected from the same source. There are treatment plants to remove iron at the wells, and the issue was addressed and resolved. Compliance is based on a running annual average. SMCLs were established to protect you against unpleasant aesthetic effects, such as color, taste, odor, and/or the staining of plumbing fixtures (e.g., tubs and sinks) and clothing when washed. Exceeding these SMCLs does not pose a health rick

8 Manganese occasionally exceeds the secondary maximum contaminant level value (SMCL) of 50 ppb in one groundwater well. The average of samples collected from this source are below the SMCL; therefore, the standard was not exceeded. SMCLs were established to protect you against unpleasant aesthetic effects, such as color, taste, odor, and/or the staining of plumbing fixtures (e.g., tubs and sinks) and clothing when washed. Exceeding these SMCLs does not pose a health risk.

9 Some groundwater wells contain 1,4-dioxane at levels greater than the notification level (NL) established by the California Department of Public Health (DPH). An NL is a health-based advisory level for unregulated contaminants in drinking water. DPH uses them to provide guidance to drinking water systems. DPH recommends taking a well out of service if the concentrations in the well exceed 35 times the NL. Our levels were well below this



East Lost Angeles District 2000 South Tubeway Avenue Commerce, CA 90040 Customer Center: (323) 722-8601 24-Hour Emergencies: (323) 263-4145

www.calwater.com

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Este informe contiene información muy importante sobre su agua potable. Tradúzcalo o hable con alguien que lo entienda bien.



About Your Water...

We're sitting down for an up close and personal interview with your water to find out what you want to know...

Interviewer: You look so clean and clear right now, but why do you sometimes contain sand or sediment?

Water: Sand or dirt can occur in groundwater, or get into water lines during repairs. The easiest thing to do is flush your faucets until the sediment disappears. That's why Cal Water occasionally flushes water from fire hydrants – to remove sediment and assure good water quality.

Interviewer: What do you say to people who say you're too hard?

Water: Well, that hurts. If I'm hard, it just means I have more minerals than my softer counterparts. Water is considered soft if its hardness is less than 75 parts per million (ppm), moderately hard at 75 to 150 ppm, hard at 150 to 300 ppm, and very hard at 300 ppm or higher. Hard water does not pose any health concerns, but some people prefer to use water softeners.

Interviewer: So what's the deal with fluoride?

Water: State law requires Cal Water to add fluoride to drinking water if public funding is available to pay for it, and it is a practice endorsed by the American Medical Association and the American Dental Association to prevent tooth decay. In this area, local water is blended with imported water that has fluoride in it. Show the table in this report to your dentist to see if he or she recommends giving your children fluoride supplements.

Interviewer: Why does water sometimes smell like chlorine?

Water: In many places water is treated to prevent the spread of germs that can cause serious illness. This can cause the water to smell like chlorine, but usually, it's fine if you refrigerate it before you drink it.

Interviewer: Water, I know you have a lot of important things to do, and I can't thank you enough for being here.

Drinking Water Source Assessment and Protection Program (DWSAPP)

By the end of 2003, Cal Water had submitted to the California Department of Public Health a DWSAPP report for each water source in the water system. The DWSAPP report identifies possible sources of contamination to aid in prioritizing cleanup and pollution prevention efforts. All reports are available for viewing or copying at our Customer Center.

The water sources in the East Los Angeles system are considered most vulnerable to contamination from gas stations, confirmed leaks, known contaminant plumes, chemical/petroleum storage, metal fabrication, and plastic producers.

We encourage customers to join us in our efforts to prevent water pollution and protect our most precious natural resource.